Adsorption transition of a polyelectrolyte on a high-dielectric charged substrate

Chi-Ho Cheng* and Pik-Yin Lai

Department of Physics and Center for Complex Systems, National Central University, Taiwan (Received 19 June 2003; published 30 December 2004)

The behavior of a polyelectrolyte adsorbed on a charged surface of high-dielectric constant is studied by both Monte Carlo simulation and analytical methods. It is found that in a low ionic strength medium, the transition is first-order with the repulsive charged surface. The surface monomer density, which is the order parameter of the adsorption transition, follows a linear relation with surface charge density. It indicates that the polyelectrolyte is compressed on the substrate without any conformational change before the desorption. Finally, a different scaling law for the layer thickness is derived and verified by simulation.

DOI: 10.1103/PhysRevE.70.061805 PACS number(s): 61.25.Hq, 82.35.Gh

Polymer adsorption on an attractive surface has drawn considerable interest due to its relation to surface effects in critical phenomena and practical importance in material science and biophysics. It is well established that the adsorption transition is continuous if its attraction on the surface is short ranged [1–4]. On the other hand, long-ranged electrostatic interactions in polyelectrolyte systems pose many challenging theoretical problems. Recently the macroion adsorption on an electrostatically attractive interface and the associated charge inversion phenomena of adsorbed polyelectrolytes acquire lots of attention [5–7].

Previous analytical approaches using the Edwards equation imposed the continuity of the monomer density across the surface and setting the monomer density to zero at the surface [8,9]. Within the framework of the self-consistent field method, both the Poisson-Boltzmann equation and the Edwards equation were solved simultaneously [10–12] with zero monomer density at the surface. These treatments, however, cannot faithfully respect the electrostatic boundary condition. As a result, the adsorption transition would still be continuous whatever the surface potential looks like.

Recently, the theoretical interest to the problem is due to its importance for multilayer polyelectrolyte adsorption [13–16]. It also raises the question of applying Poisson-Boltzmann theory to polyelectrolyte adsorption because the theory fails to capture the correlation effects.

In this paper, we study the adsorption of a single polyelectrolyte on a high-dielectric substrate in which the image charge attraction is strong. At low ionic strength, the adsorption transition occurs when the surface charges are repulsive instead of the attractive cases that were usually studied. The problem is tackled by performing Monte Carlo (MC) simulations and also by analytical methods in polymer physics taking full account of the appropriate boundary conditions. It is found that the order of the adsorption transition, the physical mechanism, and the scaling behavior are all different from those of the attractive surfaces.

A polyelectrolyte carrying positive charges is immersed in a medium $(z>0)$ of dielectric constant ϵ . At $z=0$ there is an impenetrable surface. Below the surface $(z<0)$, it is a sub-

$$
\mathcal{H} = \frac{3k_{\mathrm{B}}T}{2l_0^2} \int_0^N ds \left(\frac{\partial \vec{r}(s)}{\partial s}\right)^2 + \frac{1}{2} \int_0^N ds \int_0^N ds'
$$

\n
$$
\times \left(\Gamma \frac{\mathrm{e}^{-\kappa |\vec{r}(s) - \vec{r}(s')|}}{|\vec{r}(s) - \vec{r}(s')|} - \Gamma'(2 - \delta_{s,s'}) \frac{\mathrm{e}^{-\kappa |\vec{r}(s) - \vec{r}'(s')|}}{|\vec{r}(s) - \vec{r}'(s')|}\right)
$$

\n
$$
+ h \int_0^N ds \kappa^{-1} \mathrm{e}^{-\kappa \vec{r}(s) \cdot \hat{z}} + \omega \int_0^N ds \int_0^N ds' \delta(\vec{r}(s) - \vec{r}(s')), \tag{1}
$$

where *s* is the variable to parametrize the chain, l_0 the bare persistence length, and κ^{-1} the Debye screening length. $\vec{r}(s) = (x(s), y(s), z(s))$, $\vec{r}'(s') = (x(s'), y(s'), -z(s'))$ are the positions of the monomers and their electrostatic images, respectively. $\Gamma = q_0^2 / \epsilon$, $\Gamma' = \Gamma(\epsilon' - \epsilon) / (\epsilon' + \epsilon)$, and *h* $=4\pi q_0\sigma/(\epsilon'+\epsilon)$ are the coupling parameters governing the strength of Coulomb interactions among the monomers themselves, between the polymer and its image, and between the polymer and the charged surface, respectively. The last term in Eq. (1) represents the excluded volume interactions with $\omega > 0$ (good solvent regime) in this study. We shall focus on the case of a charge polymer in a low ionic strength medium.

The above continuum model is discretized to perform Monte Carlo simulation. The continuous curve $\vec{r}(s)$ is replaced by a chain of beads \vec{r}_i ($i=1,\ldots,N$) with hard-core excluded volume of finite radius *a*. Total lengths up to *N* =120 are employed. Units of length and energy are set to be 2*a* and $q_0^2/2\epsilon a$, respectively. Dielectric ratios ϵ'/ϵ are chosen from 2 to 12.5 (aqueous solution with a metallic substrate). Runs up to 10^9 MC steps are performed to achieve good statistics.

The adsorption layer can be characterized by the normalized monomer density $\rho(z)$. $\rho_a \equiv \rho(a)$ representing the fraction of monomers being adsorbed on the substrate is chosen *Electronic address: phcch@phy.ncu.edu.tw as an order parameter to describe the adsorption transition.

strate of dielectric constant ϵ' . Just above the substrate, there is a uniform surface charge density σ . The adsorbed polyelectrolyte always stays above the surface charge layer. Denoting the charge on a polymer segment ds by $q_0 ds$, the Hamiltonian is

FIG. 1. Monte Carlo results for the normalized monomer density at the surface ρ_a as a function of surface charge density σ (in units of $q_0 / 4a^2$) for different ϵ' / ϵ at $\kappa^{-1} = 25$. The fitted straight lines are terminated at their adsorption transition points. The vertical dashed lines are drawn as guides to the eyes. Inset: The polarization surface charge density induced by the polyelectrolyte, $|\sigma_{\rm p}|$, as a function of dielectric ratio ϵ'/ϵ . The sign of σ_p is opposite to q_0 and is negative. The solid curve is fitted from Eq. (6) with σ_{poly} $=0.118.$

 ρ_a is and ρ_a =0 characterize the adsorbed and desorbed states, respectively. In Fig. 1, ρ_a as a function of the surface charge density σ for various $\epsilon'/\epsilon > 1$ is shown. The discontinuous jump of ρ_a across the threshold indicates the transition is first order. We also verified that the energy jump (latent heat) across the transition is proportional to *N*. Similar results were obtained for larger κ^{-1} .

Furthermore, the data in Fig. 1 also indicate that ρ_a is linear in σ with the slope depending on the ratio of ϵ'/ϵ . Such a linear relation between ρ_a and σ can be understood from the electrostatic boundary conditions that the system has to satisfy. The electric potential $\phi(z)$ in the neighborhood of the *z*=0 boundary obeys

$$
-\frac{\partial \phi}{\partial z}\bigg|_{z=0^+} + \frac{\partial \phi}{\partial z}\bigg|_{z=0^-} = \frac{4\pi}{\epsilon}\bigg(\frac{2\sigma}{\epsilon'/\epsilon+1} + \sigma_{\rm p}\bigg),\tag{2}
$$

where $\sigma_{\rm p}$ is the polarization surface charge density induced by the polymer only, which depends on ϵ'/ϵ but is independent of σ in the adsorbed regime near the transition. Notice that $\sigma_{\rm p}$ in general is a complicated function since it relies on the polymer conformation. Also, if one treats the polymer as a macromolecule with a well-defined surface, its surface charge density at $z = a$ should be proportional to the monomer density ρ_a , this also applies to the electric field in the $z < 0$ region,

$$
-K \left. \frac{\partial \phi}{\partial z} \right|_{z=a^{-}} = -\frac{4\pi}{\epsilon} \rho_a, \tag{3}
$$

$$
-K \left. \frac{\partial \phi}{\partial z} \right|_{z=0^-} = -\frac{4\pi}{\epsilon'} \frac{2\epsilon'}{\epsilon' + \epsilon} \rho_a, \tag{4}
$$

where $K > 0$ is the corresponding proportional constant. Applying the continuity condition for the electric field from *z* $=0^+$ to $z=a^-$, and using Eqs. (2)–(4), one gets the linear behavior

$$
\rho_a = -\frac{2K}{\epsilon'/\epsilon - 1} \bigg(\sigma + \frac{\epsilon'/\epsilon + 1}{2} \sigma_p \bigg). \tag{5}
$$

Notice that *K* and σ_p are functions of ϵ'/ϵ . The linear behavior in Eq. (5) is confirmed by the simulation data as shown in Fig. 1, the slope decreases monotonically with ϵ'/ϵ . Substituting $\sigma=0$ into Eq. (5), we get the polarization surface charge density as a function of dielectric constant ratio,

$$
\sigma_{\mathbf{p}} = -\frac{\rho_a|_{\sigma=0}}{K} \frac{\epsilon' - \epsilon}{\epsilon' + \epsilon}.
$$
 (6)

K for different dielectric ratios is obtained from the slopes of different straight lines presented in Fig. 1. σ_p as a function of ϵ'/ϵ is then fully determined from simulation data as shown in the inset of Fig. 1. It fits very well to Eq. (6) with σ_{poly} $\equiv \rho_a |_{\sigma=0} / K = 0.118$. It suggests the conformation of the adsorbed polymer is compressed on the high-dielectric substrate. We have also checked the adsorption layer thickness is independent of the number of monomer *N*, which is consistent with the picture of a compressed state. There is no conformational change during the desorption contrary to the scaling results as predicted by Borisov *et al.* [19] for the attractive surface.

The polyelectrolyte behaves as electric blobs arranged longitudinally and lie down parallel to the surface. Increasing the attraction from the charged surface reduces the *z*-fluctuation amplitude of the chain, but the effective inplane surface charge distribution of the polyelectrolyte does not change. Hence the polarization σ_p is independent of σ . The excluded volume effect is safely ignored because it takes almost no effect in the *z* direction. The effect from selfelectrostatic interaction of the polyelectrolyte can be absorbed into the bare persistence length from l_0 to l .

Because the monomer would feel the strongest attraction from its direct image around the adsorption regime, the Γ' term in Eq. (1) is approximated by the interaction of each monomer and its corresponding image only. The residual attraction from the images of other monomers is then adsorbed into the coupling parameter Γ' from q_0 to q . The partition function is reduced to

$$
Z = \int \mathcal{D}[\vec{r}(s)] \exp \left\{ \int_0^N ds \left[-\frac{3}{2l^2} \left(\frac{\partial \vec{r}(s)}{\partial s} \right)^2 + \frac{\beta \Gamma'}{4} \frac{e^{-2\kappa \vec{r}(s) \cdot \hat{z}}}{\vec{r}(s) \cdot \hat{z}} - \beta h \kappa^{-1} e^{-\kappa \vec{r}(s) \cdot \hat{z}} \right] \right\}.
$$
 (7)

Transforming the variable from $\vec{r}(s)$ to the normalized monomer density $\rho(\vec{r}) = (1/N)\int_0^N ds \,\delta(\vec{r} - \vec{r}(s))$ by introducing an auxiliary field, then applying the ground state dominance approximation in large-*N* limit and by variational principle [17,18], one obtains the Edwards-Schrödinger equation,

where ε_0 acts as a Lagrange multiplier to enforce the constraint of the ground state wave-function normalization. The monomer density is given by $\rho(z) = |\psi(z)|^2$. Equation (8) also describes a quantum particle at its ground state moving under a combined potential of a one-dimensional (1D) screened Coulomb attraction and an almost linear potential. However, the boundary condition expressed by Eq. (5) is different from the hard-wall boundary condition $\psi|s=0$ usually employed for a quantum particle. Instead $\psi|_{s} = \sqrt{\rho_a} \neq 0$ for the present problem implies that the steric force felt by the polyelectrolyte from the charged surface should be modified [20]. Setting ψ _s=0 [8,9,21] in the polyelectrolyte adsorption problems in previous studies is not completely correct.

During the adsorption, the rodlike polyelectrolyte tends to lie down on the charged surface. The thickness of the adsorption layer is of the same order of the gyration radius in the *z* direction. At low ionic strength in which the Debye length is much greater than the layer thickness, the polyelectrolyte cannot feel the potential of length scale much larger than κ^{-1} , but only the potential barrier height is important. The original potential $V(z)$ in Eq. (8) can thus be replaced by

$$
V_{\text{mod}}(z) = \begin{cases} +\infty, & z < a \\ V(z), & a \leq z < z_{\text{br}} \\ V(z_{\text{br}}), & z \geq z_{\text{br}}, \end{cases} \tag{9}
$$

where z_{br} is chosen such that $V'(z_{\text{br}})=0$ and $V(z_{\text{br}})$ is the barrier height. In the limit of $\sigma = \kappa = 0$, analytic solution gives

$$
\psi(z) = W_{\lambda, 1/2} \left(\frac{3\beta \Gamma'}{2l^2 \lambda} z \right); \quad \varepsilon_0 = -\frac{3\beta^2 \Gamma'^2}{32l^2} \frac{1}{\lambda^2}, \quad (10)
$$

where $W_{\lambda,1/2}$ is the Whittaker's notation of the confluent hypergeometric function [22], and λ is the least value satisfying the boundary condition. The bound state exists for arbitrary ϵ'/ϵ 1. It implies the threshold surface charge density, σ_t >0 at low ionic strength.

For both $\sigma, \kappa > 0$, no exact solution exists in general

FIG. 2. (a) Simulation results for the surface charge density at the transition σ_t as a function of dielectric ratio ϵ'/ϵ in logarithmic scale at κ^{-1} $=25$. The straight lines indicate slopes of 1 and 3 as suggested in the text. (b) Simulation results for the inverse decay length α (which is proportional to the inverse layer thickness) as a function of σ (in units of $q_0/4a^2$) for $\epsilon'/\epsilon=12.5$, $\kappa^{-1}=25$. The straight line is a linear fit. σ _t=0.102 in this case. α is obtained from exponential fitting to the tail of corresponding density profile.

but one can analyze it around the transition. Near the surface, the image charge attraction dominates over the surface charge repulsion and hence the binding energy is approximated by the $\sigma=0$ case in Eq. (10). The polyelectrolyte undergoes a desorption transition when the binding energy meets the barrier height $V(z_{\text{br}})$. After some algebra, we have $\sigma_t \sim (\epsilon'/\epsilon - 1)$ for $\epsilon'/\epsilon \ge 1$ and $\sigma_t \sim (\epsilon'/\epsilon - 1)^3$ for $\epsilon'/\epsilon \ge 1$. This analytic result is consistent with our simulation data as shown in Fig. 2(a).

An approximate solution for the density profile $\rho(z)$ for the σ > 0 case can be obtained by variational method with trial wave function,

$$
\psi(z) = \sqrt{\rho_a} \left[1 + \mu \alpha(z - a) \right] e^{-(1/2)\alpha(z - a)},\tag{11}
$$

where α^{-1} is the decay length. μ is positive because the trial wave function is restricted to be nodeless. α and μ are not independent but related via the wave-function normalization condition. The inverse decay length is calculated to be

$$
\alpha = 3\beta \Gamma'/2l^2 + \rho_a,\tag{12}
$$

where the leading term is independent of σ . Near the transition, the decay length and hence the thickness of the adsorption layer increase and remain finite. From Eqs. (5) and (12), we get the scaling behavior

$$
\alpha - \alpha_{t} \sim (\sigma_{t} - \sigma) \quad \text{for } 0 < \sigma < \sigma_{t}, \tag{13}
$$

where α_t is the threshold inverse layer thickness at $\sigma = \sigma_t^-$. The variation of α as a function of σ obtained from the simulations is shown in Fig. 2(b) which can be well fitted to a linear relation consistent with Eq. (13). On the other hand, for the case of adsorption onto an attractive charged surface $(\sigma < 0)$ with substrate of $\epsilon'/\epsilon \leq 1$ (e.g., DNA in aqueous solution adsorbed onto a charged lipid membrane), the asymptotic solution to Eq. (8) reproduces the usual scaling $\sigma \sim |\sigma|^{1/3}$ and is a continuous adsorption transition [13], and the thickness swells to infinity as the polyelectrolyte is desorbed.

A strongly charged polyelectrolyte immersed in a salt solution will attract oppositely charged ions to condense until its effective charge density reaches the Manning threshold [23]. This means that one can just renormalize q_0 in our system to $2ea/l_B$ if q_0 is larger than $2ea/l_B$ (l_B is the Bjerrum length). Similarly, the strongly charged surface of bare charge density larger than $\kappa/(\pi l_B)$ is just renormalized back to $\kappa/(\pi l_B)$ [24]. The Gouy-Chapman length, calculated from the nonlinear Poisson-Boltzmann theory, is of the order $(l_B \sigma)^{-1}$, which is very large around the transition. However, the nonlinear Poisson-Boltzmann potential [25] near the substrate is given by

$$
\phi_{\rm PB}(z) = \frac{2k_{\rm B}T}{e} \ln \frac{1 + \gamma e^{-\kappa z}}{1 - \gamma e^{-\kappa z}} = \phi_0 - \frac{2ez}{\epsilon_0 l_{\rm B} \lambda_{\rm GC}} + O(z^2),\tag{14}
$$

where λ_{GC} is the Gouy-Chapman length and $\gamma = \sqrt{\kappa^2 \lambda_{\text{GC}}^2 + 1}$ $-\kappa\lambda_{\text{GC}}$. Notice that the linear term in *z* is proportional to $\lambda_{\text{GC}}^{-1} \propto \sigma$ and is identical to the linear term as expanded from the Debye-Hückel potential. It is not surprising since both Poisson-Boltzmann and Debye-Hückel potentials share the same boundary condition. When the adsorbed polyelectrolyte layer thickness is of one to two monomer size [as seen from α^{-1} ~ 1 in Fig. 2(b)], the surface potential felt by the polyelectrolyte should be linear. Physically speaking, it does not matter whether the potential is Poisson-Boltzmann or Debye-Hückel, or even the linear one if the layer thickness is much smaller than λ_{GC} and κ^{-1} . The potential near the surface is determined from the boundary condition. Unlike the case of an attractive surface [26], the effect from large Guoy-Chapman length near the transition is irrelevant in the highdielectric case.

Our results on the single polyelectrolyte adsorption may provide a starting point to study the charge inversion and multilayer adsorption [27]. At low ionic strength, polyelectrolytes are adsorbed in a multilayer structure because of strong Coulomb repulsion. Each layer is composed of parallel 1D Wigner crystal [28]. The upper bound of the multilayer thickness is $z_{\text{br}} \sim \sigma^{-1/2} (\epsilon'/\epsilon - 1)^{1/2}$. It suggests we can easily adjust a single layer adsorbed onto a highdielectric substrate by tuning the surface charge density. Rigorous treatment based on this physical picture will be elaborated elsewhere [29].

In conclusion, the adsorption transition of a single polyelectrolyte on a high-dielectric substrate is first order since the polyelectrolyte needs to overcome a binding energy from its image charge. Because of the strong Coulomb attraction as compared to the linear repulsive potential near the surface, the polyelectrolyte is compressed without any conformational change before the desorption. A scaling law for the adsorption layer thickness is also derived and verified by simulation.

C.H.C. would like to thank M. Rubinstein for helpful comments. This work was supported by the National Science Council of the Republic of China under Grant Nos. NSC91- 2816-M-008-0009-6 (C.H.C.) and NSC92-2112-M-008-051 (P.Y.L.).

- [1] P. G. de Gennes, Macromolecules **14**, 1637 (1981).
- [2] P. G. de Gennes and P. Pincus, J. Phys. (France) Lett. **44** L241 (1983).
- [3] E. Eisenreigler, K. Kremer, and K. Binder, J. Chem. Phys. **77**, 6291 (1982).
- [4] P. Y. Lai, Phys. Rev. E **49**, 5420 (1994).
- [5] W. M. Gelbart, R. F. Bruinsma, P. A. Pincus, and V. A. Parsegian, Phys. Today **53**(9), 38 (2000).
- [6] *Electrostatic Effects in Soft Matter and Biophysics*, edited by C. Holm, P. Kékicheff, and R. Podgornik (Kluwer, Dordrecht, 2001).
- [7] A. Y. Grosberg, T. T. Nguyen, and B. I. Shklovskii, Rev. Mod. Phys. **74**, 329 (2002).
- [8] F. W. Wiegel, J. Phys. A **10**, 299 (1977).
- [9] M. Muthukumar, J. Chem. Phys. **86**, 7230 (1987).
- [10] R. Varoqui, A. Johner, and A. Elaissari, J. Chem. Phys. **94**, 6873 (1991).
- [11] I. Borukhov, D. Andelman, and H. Orland, Europhys. Lett. **32**, 499 (1995).
- [12] X. Chatellier and J. F. Joanny, J. Phys. II **6**, 1669 (1996).
- [13] J. F. Joanny, Eur. Phys. J. B **9**, 117 (1999).
- [14] R. R. Netz and J. F. Joanny, Macromolecules **32**, 9013 (1999).
- [15] F. J. Solis and M. O. de la Cruz, J. Chem. Phys. **110**, 11 517 (1999).
- [16] T. T. Nguyen, A. Y. Grosberg, and B. I. Shklovskii, Phys. Rev. Lett. **85**, 1568 (2000).
- [17] M. Doi and S. F. Edwards, *The Theory of Polymer Dynamics* (Oxford University, New York, 1986).
- [18] T. Garel, H. Orland, and E. Pitard, in *Spin Glasses and Random Fields*, edited by A. P. Young (World Scientific, Singapore 1998).
- [19] O. V. Borisov, E. B. Zhulina, and T. M. Birshtein, J. Phys. II **4**, 913 (1994).
- [20] C. H. Cheng (unpublished).
- [21] R. R. Netz and D. Andelman, Phys. Rep. **380**, 1 (2003).
- [22] R. Loudon, Am. J. Phys. **27**, 649 (1959).
- [23] G. S. Manning, J. Chem. Phys. **51**, 924 (1969).
- [24] L. Bocquet, E. Trizac, and M. Aubouy, J. Chem. Phys. **117**, 8138 (2002).
- [25] J. N. Israelachvili, *Intermolecular and Surface Forces* (Academic, London, 1985).
- [26] A. V. Dobrynin, A. Deshkovski, and M. Rubinstein, Phys. Rev. Lett. **84**, 3101 (2000).
- [27] A. V. Dobrynin, A. Deshkovski, and M. Rubinstein, Macromolecules **34**, 3421 (2001).
- [28] B. I. Shklovskii, Phys. Rev. Lett. **82**, 3268 (1999).
- [29] C. H. Cheng and P. Y. Lai, e-print cond-mat/0403722.